

Synthesis and Antibacterial Activity of 5-Nitro-2-furfurylidene *p*-Sulfonyl- and *p*-Sulfamoylbenzoylhydrazides VII

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Abstract □ Thirteen 5-nitro-2-furfurylidene *p*-sulfonyl- and *p*-sulfamoylbenzoylhydrazides were synthesized. All compounds prepared were tested against 10 Gram-positive and Gram-negative bacteria and exhibited antibacterial activity.

Keyphrases □ 5-Nitro-2-furfurylidene *p*-sulfonyl- and *p*-sulfamoylbenzoylhydrazides—synthesis, antibacterial activity □ Antibacterial activity—evaluation of 13 5-nitro-2-furfurylidene *p*-sulfonyl- and *p*-sulfamoylbenzoylhydrazides

In continuation of studies on the preparation of new 5-nitro-2-furfurylidene derivatives (1) and on

the synthesis and pharmacological activity of compounds related to alkylsulfonyl-, arylsulfonyl-, and sulfamoylbenzoylhydrazides (2–5), a series of 5-nitro-2-furfurylidene benzoylhydrazides having sulfonyl or sulfamoyl groups in the *para*-position was synthesized. *p*-Sulfonyl- and *p*-sulfamoylbenzoylhydrazides were prepared by interaction of hydrazine hydrate and an appropriate ethyl ester (3, 4). The new benzoylhydrazides prepared are summarized in Table I.

Benzoylhydrazides were allowed to react with 5-nitro-2-furaldehyde to give the 5-nitro-2-furfurylidene derivatives (Scheme I). The 5-nitro-2-furfuryli-

Table I—*p*-Sulfamoylbenzoylhydrazides

R	Melting Point	Yield, %	Formula	<i>p</i> -RNHSO ₂ C ₆ H ₄ —CONHNH ₂	
				Analysis, %	
				Calc.	Found
CH ₃	143°	65	C ₈ H ₁₁ N ₃ O ₃ S	C 41.92	41.88
				H 4.80	4.71
C ₂ H ₅	150–152°	50	C ₉ H ₁₃ N ₃ O ₃ S	C 44.44	44.54
				H 5.34	5.28
(CH ₃) ₂ CH	168–169°	80	C ₁₀ H ₁₅ N ₃ O ₃ S	C 46.69	46.73
				H 5.83	5.81
C ₆ H ₅	220–222°	57	C ₁₃ H ₁₃ N ₃ O ₃ S	C 53.60	53.66
				H 4.46	4.52

Table II—5-Nitro-2-furfurylidene *p*-Sulfonyl- and *p*-Sulfamoylbenzoylhydrazides


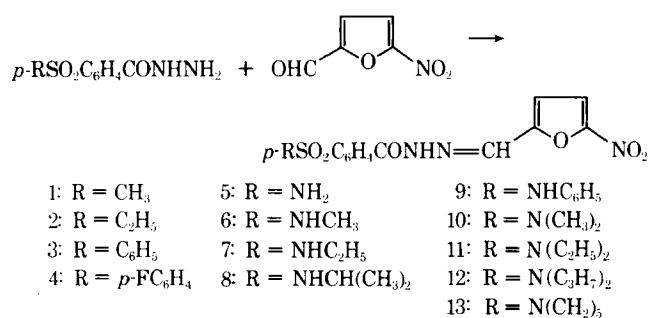
Compound	R	Melting Point	Yield, %	Formula	<i>p</i> -RSO ₂ C ₆ H ₄ —CONHN=CH— 	
					Analysis, %	
					Calc.	Found
1	CH ₃	265°	82	C ₁₃ H ₁₁ N ₃ O ₆ S	C 46.29	46.25
					H 3.26	3.26
2	C ₂ H ₅	240°	85	C ₁₄ H ₁₃ N ₃ O ₆ S	C 47.86	48.02
					H 3.70	3.75
3	C ₆ H ₅	237°	92	C ₁₈ H ₁₃ N ₃ O ₆ S	C 54.13	54.14
					H 3.25	3.33
4	<i>p</i> -FC ₆ H ₄	238–243°	90	C ₁₈ H ₁₂ FN ₃ O ₆ S	C 51.79	51.82
					H 2.87	2.77
5	NH ₂	285°	83	C ₁₂ H ₁₀ N ₄ O ₆ S	C 42.60	42.61
					H 2.95	3.09
6	CH ₃ NH	297°	79	C ₁₃ H ₁₂ N ₄ O ₆ S	C 44.31	44.30
					H 3.40	3.44
7	C ₂ H ₅ NH	258°	81	C ₁₄ H ₁₄ N ₄ O ₆ S	C 45.90	45.99
					H 3.82	3.82
8	(CH ₃) ₂ CHNH	254–259°	77	C ₁₅ H ₁₆ N ₄ O ₆ S	C 47.36	47.40
					H 4.21	4.40
9	C ₆ H ₅ NH	275°	75	C ₁₈ H ₁₄ N ₄ O ₆ S	C 52.17	52.17
					H 3.38	3.44
10	(CH ₃) ₂ N	227°	89	C ₁₄ H ₁₄ N ₄ O ₆ S	C 45.90	45.80
					H 3.82	3.70
11	(C ₂ H ₅) ₂ N	243°	93	C ₁₆ H ₁₈ N ₄ O ₆ S	C 48.73	48.69
					H 4.56	4.51
12	(C ₃ H ₇) ₂ N	210°	89	C ₁₈ H ₂₂ N ₄ O ₆ S	C 51.18	51.18
					H 5.21	5.23
13	(CH ₃) ₅ N	250°	79	C ₁₇ H ₁₈ N ₄ O ₆ S	C 48.98	49.11
					H 4.54	4.50

Table III—Antibacterial Activity

Compound	Zone of Inhibition, Average Size, mm					
	<i>Staph. aureus</i>	<i>Staph. albus</i>	<i>K. pneumoniae</i>	<i>S. faecalis</i>	<i>B. cereus</i>	<i>B. subtilis</i>
1	—	9.9	—	—	—	12.4
2	13	13.8	—	—	—	13
3	11	12	10.1	—	—	12
4	9.11	9.9	—	—	10.8	11.5
5	—	—	—	—	—	—
6	10.6	—	—	—	—	—
7	11.2	10.3	—	—	—	10.7
8	11.1	11.3	11.1 ^a	—	—	10.7
9	10.8	11.8	—	—	11	11.4
10	9.5	11.3	—	—	—	11.3
11	13.3 ^a	12.3	—	—	11.8 ^a	14.3
12	—	11.9	—	—	—	10.4
13	11.2	13.9	—	—	11.8	13.8
Furazolidone	23.3	22.7	20.2	13.1	23.4	25.5

^a Hazy zones.

Scheme 1

EXPERIMENTAL^{2,3}

***para*-Substituted Benzoylhydrazides**—A *para*-substituted benzoyl ethyl ester (0.02 mole) was dissolved in 25 ml of ethanol, and then 0.025 mole of 99% hydrazine hydrate was added to the hot solution. After refluxing the reaction mixture for 1 hr, the solvent was distilled off and the residue was crystallized from aqueous ethanol (Table I).

5-Nitro-2-furfurylidene *p*-Sulfonyl- and *p*-Sulfamoylbenzoylhydrazides—To a hot solution of 0.01 mole of the appropriate benzoylhydrazide in 25 ml of alcohol, a solution of 0.01 mole of 5-nitro-2-furaldehyde in 5 ml of alcohol was added. After standing for 3 hr at room temperature, the crystalline mass was filtered and recrystallized from alcohol (Table II).

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dene *p*-sulfonyl and *p*-sulfamoyl derivatives are summarized in Table II.

PHARMACOLOGY

The compounds listed in Table II were tested against 10 Gram-positive and Gram-negative bacteria. Furazolidone was used as a control. The compounds were dissolved in acetone and diluted with water or phosphate buffer (pH 8) to a final concentration of 250 μ g/ml. The solvent mixture was acetone-water (0.5:1.5) for Compounds 1, 11, and 17; acetone-buffer (1:1) for Compounds 5 and 6; and acetone-water (1:1) for all other compounds. Paper disks of 9-mm diameter were immersed in the solutions and put on the inoculated surface of penicillin assay seed agar.

All compounds and furazolidone were inactive against *Bordetella bronchiseptica* (ATCC 4617), *Sarcina lutea* (ATCC 341a), and *Proteus vulgaris*¹. None of the compounds prepared showed significant activity against *Escherichia coli* (0128) at the test concentration. The antibacterial activities of other compounds against *Staphylococcus aureus* (ATCC 6538-P), *Staphylococcus albus* (ATCC 12228), *Klebsiella pneumoniae* (ATCC 10031), *Streptococcus faecalis* (ATCC 8043), *Bacillus cereus* (ATCC 1178), and *Bacillus subtilis* (NCTC 3610) are reported in Table III.

¹ Obtained from S. S. Pfizer Laboratories, Tehran, Iran.² Melting points were taken on a Kofler hot stage microscope and are uncorrected. The IR spectra were recorded using a Leitz spectrograph. NMR spectra were obtained with a Varian A-60A instrument. The mass spectra were recorded on a Varian Mat 111 spectrograph.³ All compounds were subjected to IR, NMR, and mass spectroscopy and the results were as expected.